(c)2001-2005 Mineral Data Publishing, version 1

Crystal Data: Orthorhombic. Point Group: n.d. Crystallites are lathlike or prismatic, to  $4 \mu m$ , in microconcretions.

Physical Properties: Hardness = n.d. D(meas.) = n.d. D(calc.) = 3.43

**Optical Properties:** Semitransparent. *Color:* Turquoise green, green, blue. *Luster:* Dull to waxy.

Optical Class: Biaxial. Orientation:  $X \parallel$  elongation;  $Z \perp$  elongation.  $\alpha = 1.57$   $\beta = [1.77]$   $\gamma = 1.95$  2V(meas.) = n.d.

Cell Data: Space Group: n.d. a=5.35 (by analogy to synthetic  $Cu(C_2O_4) \cdot 0.1H_2O$ . b=5.63 c=2.56 Z=1

**X-ray Powder Pattern:** Mooloo Station, Australia. 3.88 (100), 2.50 (30), 1.753 (30), 2.31 (25), 1.787 (25), 2.14 (20), 2.33 (18)

Chemistry:

$$\begin{array}{cc} & & (1) \\ \mathrm{SiO}_2 & [6.35] \\ \mathrm{CuO} & [46.7] \\ \mathrm{C}_2\mathrm{O}_3 & 42.3 \\ \mathrm{H}_2\mathrm{O} & 4.65 \\ \hline \mathrm{Total} & [100.0] \\ \end{array}$$

(1) Mooloo Station, Australia; CHN analyzer gave C 14.10% and H 0.52% and a positive test for oxalate was obtained; calculating CuO for stoichiometry and assuming  $SiO_2$  by difference, corresponds to  $Cu(C_2O_4) \cdot 0.44H_2O$ .

Occurrence: On an outcrop of quartz, thought to have formed by reaction between bird guano and soluble secondary copper minerals (Mooloo Station, Australia); in a mine shaft near tree roots (Sainte-Marie-aux-Mines, France).

**Association:** Opaline silica, sampleite, libethenite, brochantite, antlerite, atacamite, whewellite, chalcopyrite, digenite, covellite, gypsum, barite, jarosite (Mooloo Station, Australia).

**Distribution:** From Mooloo Station, 12 km east of the homestead, Murchison, Western Australia. At Sainte-Marie-aux-Mines, Haut-Rhin, France.

Name: For the original locality on Mooloo Station, Australia.

Type Material: Western Australian Museum, Perth, Australia, MDC6738.

**References:** (1) Clarke, R.M. and I.R. Williams (1986) Moolooite, a naturally occurring hydrated copper oxalate from Western Australia. Mineral. Mag., 50, 295–298. (2) (1987) Amer. Mineral., 72, 1025–1026 (abs. ref. 1).